

## 3D Structure of Nanomaterials under Realistic Conditions

**Prof. Sara Bals**

**Dr. Wiebke Albrecht, Dr. Eva Bladt et al.**

*EMAT, University of Antwerp, Belgium*

Electron tomography has evolved into a powerful tool to investigate a broad variety of nanomaterials, also at the atomic scale. However, 3D characterisation by transmission electron microscopy (TEM) is conventionally performed in ultra-high vacuum and at room temperature. To fully understand the connection between the 3D structure and properties under realistic conditions, innovative methodologies are required to track the fast 3D changes of nanomaterials that occur in different thermal and gaseous environments.

For example, to investigate the thermal stability of metallic nanoparticles, a combination of a tomographic heating holder with fast tilt series acquisition has been used. First, we investigated the morphological evolution of a Au nanostars at elevated temperatures and quantified local volume changes [1]. Figure 1 shows the original structures in blue with the final structures after heating in red. Moreover, using the same approach, we showed that Au/Pd octopods displayed an improved thermal stability in comparison to pure Au octopods, preserving their complete morphology up to 450 °C when heated in vacuum, see figure 2 [2]. Next, we performed an accurate 3D compositional quantification based on fast tomography. We hereby measured the 3D elemental redistribution dynamics of individual bimetallic nanoparticles upon heating. We conclude that for a given composition, the shape of the nanoparticle does not influence the alloying process significantly. Based on our analysis, it is clear that interdiffusion of metals at the nanoscale is more complex than predicted by simple Fickian diffusion and that other factors such as surface diffusion need to be taken into account [3].

To quantify nanoparticle shape dynamics in a gaseous environment in 3D, HAADFSTEM images served as an input for atom counting procedures followed by 3D relaxation of the structure. In this manner, we characterized shape changes of a Pt nanoparticle in a gaseous environment (Figure 3). The conditions have been varied from vacuum (Figure 3.a) to a 1 bar H<sub>2</sub> flow (Figure 3.b), followed by a 1 bar O<sub>2</sub> environment (Figure 3.c). To investigate the behaviour during cycling, we repeated the experiment several times using the same particle (Figure 3.d&e). We clearly observe morphology changes and we were even able to quantify the occurrence of the different surface facets [4].

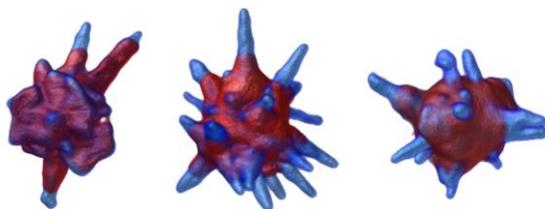


Figure 1

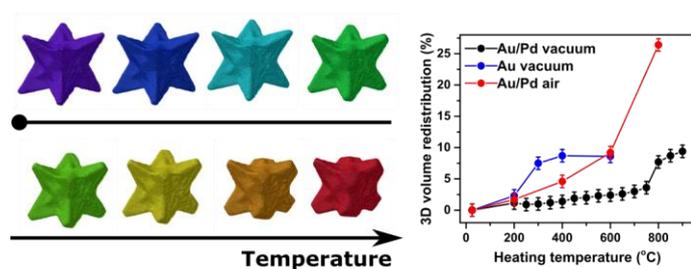


Figure 2

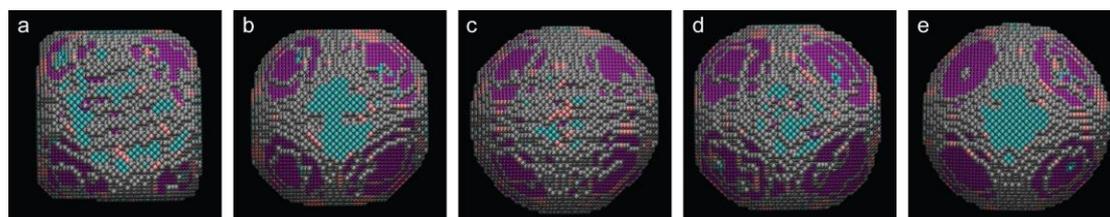


Figure 3

References:

1. H. Vanrompay, E. Bladt, W. Albrecht, A. Béché, M. Zakhozheva, A. Sánchez-Iglesias, L. M. Liz-Marzán, S. Bals, *Nanoscale* 10 (2018) 22792
2. A. Skorikov, W. Albrecht, E. Bladt, X. Xie, J. E. S. van der Hoeven, A. van Blaaderen, S. Van Aert, S. Bals *ACS Nano* accepted
3. W. Albrecht, E. Bladt, H. Vanrompay, J.D. Smith, S.E. Skrabalak, S. Bals, *ACS Nano* 13 6522 (2019)
4. T. Altantzis, I. Lobato, A. De Backer, A. Béché, Y. Zhang, S. Basak, M. Porcu, Q. Xu, A. Sánchez-Iglesias, L. M. Liz-Marzán, G. Van Tendeloo, S. Van Aert, S. Bals, *Nano Letters* 19 (2019) 477

[www.premc.org/conferences](http://www.premc.org/conferences)  
[annic2019@premc.org](mailto:annic2019@premc.org)